

GANSKAU, N.K.; POYUROVSKIY, M.Ye.

Transistorized d.c.converter with magnetic stabilization.

Priborostroenie no.1:13-15 Ja '61.

(MIRA 14:1)

(Electric current converters)

MARGULIS, L.I.; GANSKIY, V.A.; GOBERMAN, M.D., otv.red.; PEVZNER, A.S.,
zav.red.izd-va; OSENKO, L.M., tekhn.red.

[Uniform time and pay standards for construction, assembly, and
repair operations in 1960] Edinye normy i rastsenki na stroi-
tel'nye, montazhnye i remontno-stroitel'nye raboty, 1960 g.
Moskva, Gos.izd-vo lit-ry po stroit., arkhitekt. i stroit.materia-
lam. Sbornik 33. [Assembling the equipment for metallurgical
plants] Montazh metallurgicheskogo oborudovaniia. No.1. [Equipment
for blast-furnace plants] Oborudovanie domennykh tsekhov. 1960.
63 p. (MIRA 13:6)

1. Russia (1923- U.S.S.R.) Gosudarstvennyy komitet po delam stroi-
tel'stva. 2. Tsentral'noye normativno-issledovatel'skoye byuro
Ministerstva stroitel'stva RSFSR (for Margulis, Ganskiy).
(Wages) (Blast furnaces)

GANSON, P.P.; ZENKEVICH, N.L.; SERGEYEV, I.V.; UDINTSEV, G.B.

Maximum depths of the ocean. Priroda 48 no.6:84-88 Je '59.
(MIRA 12:5)

1. Institut okeanologii AN SSSR, Moskva.
(Deep-sea sounding)

BARK, L.S.; GANSON, P.P.; MEYSTER, N.A.; DOBROVOL'SKIY, A.D., prof.,
otv.red.: KORKINA, A.I., tekhn.red.

[Tables of the speed of sound in sea water] Tablitsy skorosti
zvuka v morskoi vode. Moskva, Vychislitel'nyi tsentr AN SSSR,
1961. 180 p. (MIRA 14:6)
(Sound--Speed)

CANSON, P.P.

Some details of the thermal structure of the surface layer of water
in the ocean. Trudy Inst.ocean. 45:98-101 '61. (MIRA 15:2)
(Pacific Ocean--Ocean temperature)

S/213/62/002/001/002/002
1068/1242

AUTOR: Ganson, P. P.
TITLE: Sea acoustics — a new branch of oceanology
PERIODICAL: Okeanologiya, v. 2, no. 1, 1962, 151-153

TEXT: Sea acoustics, the science of submarine sounds, is a powerful tool in the age of the atomic submarine and underwater missile launching. Proper application of the science depends on very accurate determination of the speed of sound and the acoustic properties of water. These properties should be determined on a regional basis by the integration of all available data and particularly the vertical distribution of salinity and temperature, structure and morphology of the ocean floor, and the distribution of plankton. ✓

Card 1/1

AUTHOR: Ganson, P. P.

50-58-4-11/26

TITLE: On the Computation Formulae of the Propagation Velocity of Sound in Sea Water (0 formulakh dlya rascheta skorosti rasprostraneniya zvuka v morskoy vode)

PERIODICAL: Meteorologiya i Gidrologiya, 1958, Nr 4, pp 30-34 (USSR)

ABSTRACT: The exact knowledge of the sound velocity in sea water is of importance for the solution of practical problems in case of application of hydro-acoustic devices. The velocity is ascertained by 2 ways: 1) Directly measured by apparatus and 2) computed by means of formulae, which express the dependence on temperature, salt content, and hydrostatic pressure. The advantages and disadvantages of these two methods are discussed. Also in case of presence of an apparatus the formula must be used for the computation:

$$c = \sqrt{\frac{\gamma}{\rho\beta}} \quad (1),$$
 whereby γ denotes the relation of the specific heat capacities of the sea water, ρ - the average density, and β - the coefficient of the isothermal compressibility of the sea water. According to this formula tables (references 1,2,5,6,8,10) are computed. Out of the

Card 1/3

On the Computation Formulae of the Propagation Velocity of Sound in Sea Water 50-58-4-11/26

latest works on direct measurements (references 4, 11) can be seen that the velocities, computed on the basis of tables and nomograms, which base upon a theoretical formula, are in the average by 3-4 m/sec. lower than the effective values. The coefficient β was taken too high. There are some more empirical formulae (references 4, 7, 9, 12). In the following the various formulae are compared with each other. Though the results of the latter differ from each other, the question of the choice of a formula, which gives the most accurate results, cannot be solved. The author used 80 directly measured velocities as specimens (Etalon), for which simultaneous measurements of temperature and salt content were present. Of this data curves of the measured and of the after various formulae computed sound velocity were constructed (figure 1). From the comparison the accuracy of each formula could be estimated. From this comparison the author deduces the following conclusions: 1) To be able to compute the sound velocity by the theoretical formula (1) the coefficient β must be determined by means of laboratory experiments. 2) All here discussed empirical formulae do not guarantee the accuracy which at present is required.

Card 2/3

On the Computation Formulae of the Propagation Velocity of Sound in Sea Water 50-58-4-11/26

3) For the determination of the sound velocity from the the temperature and the salt content of sea water the nomogram or the tables by Del'Grosso (reference 4) have to be applied. 4) In his formula the coefficients in the 2nd and 3rd term are to be modified (being given in this new form). 5) Because of the troublesomeness of the computation by formulae a special computation device should be constructed.

There are 3 figures, 1 table, and 12 references, 2 of which are Soviet.

AVAILABLE: Library of Congress

1. Underwater sound - Propagation Velocity
2. Underwater sound - Velocity
3. Mathematics - Applications

Card 3/3

CANSTER, F.I.

Prevent accidents. Avtomobilist 1:82-86 '61.
(Drinking and traffic accidents)

(MIRA 15:1)

GANSZER W.

ABST. JOUR. : REKHAIE., No. 20 1959, No. 72399

AUTHOR : Bankowski, C.; GANSZER, W.

ISSY. :

TITLE : Content of Tannins in Polish Species of
Lam. viscaria L.

ORIG. PUB. : Acta polon. pharmac.. 1958, 15, No 6,
481-483

ABSTRACT : A description of locations where the plants
occur; of conditions of their preliminary treatment;
results of analyses of specimens. It is proposed to utilize
the plants as pharmaceutical raw material. -- D. Yakesh.

CARD:

46

GANSZER, W., SZTAJEROWA, M.

Severe acute uremia with convulsions following allergy to sulfonamides. Polski tygod, lek. 5:8, 20 Feb. 50. p. 284-90

1. Of the Surgical Department of the Hospital imienia Przemyslaw II. in Kalisz. (Director of the Hospital--Karol Piotrowski, M. D.; Head of the Surgical Department--Lodz. Ganszer, M. D.; Head of Children's Department--L. Fiszler, M. D.).

CIML 19, 5, Nov., 1950

GANSZER, W.

Observations on postoperative syndrome of pallor and hyperthermia
in children. Polski prsegl. chir. 24 no. 1:1-7 Jan-Feb 1952.

(CJML 23:3)

1. Of the Surgical Department (Head--Wl. Ganszer, M. D.) of Hospital
imienia Przemyslaw II (Director -- Karol Piotrowski, M. D.) in
Kalisz.

GANSZER, W.; GLADYSZ, B.

Observations on application of contrast media in cases of so-called hemorrhagic cysts in the supracravicular region. Polski przezl. radiol. 17 no.1:37-46 Jan-Mar 1953. (CJML 24:5)

1. Of the Hospital imienia Przemyslaw II in Kalisz.

GANSZER, Włodzimierz

Treatment of spinal fractures with neurological lesions. Chir. narz.
ruchu 22 no.4:419-421 1957.

1. Z Oddziału chirurgicznego Szpitala Miejskiego w Kaliszu. Ordynator:
dr W. Ganszer. Kalisz, Szpital Miejski.

(SPINE, fractures

causing neurol. disord., ther., conservative & surg. (Pol))

(CENTRAL NERVOUS SYSTEM, dis.

caused by spinal fract., ther., conservative & surg. (Pol))

GANSZER, W.

POLAND/Morphology of Man and Animals (Normal and Pathologic).
Digestive System.

S-2

Abs Jour : Ref Zhur - Biol., No 4, 1958, 17024

Author : Ganszer, W.

Inst :

Title : A Case of Rectum Duplex.

Orig Pub : Polski przegl, chirurg., 1957, 29, No 5, 495-498

Abstract : No abstract.

Card 1/1

GANSZER, Włodzimierz

Extensive diverticulosis of the jejunum complicated by primary cancer of one of diverticula. Polski przeł. chir. 29 no.7:705-707 July 57.

1. Z oddziału Chirurgicznego Szpitala Miejskiego w Kaliszu. Ordynator: dr. W. Ganszer.

(JEJUNUM, diverticula,

extensive with primary cancer of one diverticulum (Pol))

POLAND

GANSZES, Wladzimierz, Dr. med., Ordynator of the Surgical Division (Oddzial Chirurgiczny) of the Municipal Hospital (Szpital Miejski) in Kalisz (Director: Dr. med. L. KOWACKI)

"Post-traumatic Liver Cyst. Case Report."

Warsaw, Polski Tygodnik Lekarski, Vol 18, No 2, 7 Jan 63, pp 62-64.

Abstract: [Author's English summary modified] Symptoms, diagnostic procedure, and successful treatment of a rare complication connected with an injury to the liver following trauma is described. There are two (2) English references.

1/1

POLAND

GANSZER, Wladzimir, Dr. med., "Ordynator" of the First Surgical Division (Oddzial Chirurgiczny I), Municipal Hospital (Szpital Miejski) in Kalisz.

"Arterious-Venous Fistula on the Face. Case Report."

Warsaw, Polski Tygodnik Lekarski, Vol 18, No 3, 14 Jan 63, pp 109-110

Abstract: [Author's English summary] A case of acquired arterious-venous fistula localized on the head (frontal vessels) is reported. Surgical treatment (excision) was successful. There are 7 references, of which 6 are Polish and one English.

1/1

GANSZER, Włodzimierz , dr.

Fracture of the carpal scaphoid bone. Chir. narząd. ruchu ortop.
Pol. 28 no.7:991-994 '63

1. Z Oddziału Chirurgicznego Szpitala Miejskiego w Kaliszu
(Ordynator: dr. W. Ganszer).

GANSZKY, Karoly

"Chemical processes in high-voltage electric discharges and their industrial applications" by Theodor Rummel. Reviewed by Karoly Ganszky. Elektrotechnika 51 no.7/9:427-428 '58.

GANSZKY, Karoly, okleveles villamosmérnök, adjunktus

External characteristic curves of six-phase rectifiers during the passive operation of the suction-choke coil. Elektrotechnika 57 no.7:269-279 J1 '64.

1. Chief, Division of Industrial Electronics, Electric Power Industry Research Institute, Budapest, and Chair of Automation, Budapest Technical University, Budapest, V., Jozsef Attila u.24.

GANSZKY, Karoly, okleveles villamosmernok, adjunktus

Current conducting conditions of the six-phase rectifier circuit with interphase transformer during the active operation of the transformer. Elektrotechnika 57 no.11/12: 537-547 N-D '64.

1. Scientific Division Chief, Research Institute of Electric Industry, and Chair of Automation, Budapest Technical University.

GANTAMIROV, Yu.S. (Groznyy)

Arrangement of pumps at the pumping station of a circulatory water-
supply system. Vod. i san. tekhn. no.1:15 Ja '61. (MIRA 14:9)
(Pumping stations)

GANTAMIROV, Yu.S., inzh. (g. Groznyy)

From the experience of designing the return water-supply systems
of process installations of petroleum refineries. Vod. i san.
tekhn. no.10:21-22 0 '65. (MIRA 18:11)

GANTAR, J.

The Arnes Hollow. p. 149; Slovenska akademija znanosti in umetnosti.
Institut za raziskovanje krasa. POROCILA. ACTA CARSOLOGICA. Ljubljana;
Vol. 1, 1955

SOURCE: East European Accessions List (EEAL), Library of Congress,
Vol. 5, No. 12, December 1956.

CZECHOSLOVAKIA

GANTCHEV, G.; KOITCHEVA, V.; [Affiliation not given].

"Functional State and Regulation Mechanisms of the Motor Analyzer in Postural Activity of Man."

Prague, Activitas Nervosa Superior, Vol 8, No 2, Jun 66, pp 160-165

Abstract [Authors' English summary modified]: Hand tremor components differing in frequency and amplitude are described. Frequency of medium amplitude oscillations was 1.2 to 2.2 cycles, at low oscillations 2.5 cycles. Visual feedback in the maintaining of a definite position has the character of negative feed-back as a stabilizing process. Low frequency components of the tremor are of supraspinal origin. Effect of the tremorogram of the functional changes in the peripheral muscular reception shows that high-frequency components are associated with the spinal regulation mechanism of postural activity. In the state of fatigue a delay in the nerve impulses occurs probably in the synapses. Postural activity was studied in relatively simple interrelations of the center and the periphery because, as proved by electromyography, the extensors play the main part in maintaining posture. 6 Figures, 1 Table, 4 Western, 5 Russian, 2 East German references. Article 1/1 is in French.

frequency: 1.2-2.2 c/sec); and 3) very weak, high frequency oscillations (~2.4 c/sec). An extended discussion searches for the mechanism(s) explaining the observed oscillations. There are 2 Bulgarian, 1 Soviet, and 2 Western references. (Manuscript received, 17 Jan 66.)

NYERES, T.: NYERES, T.: NYERES, T.

Newer data on the chemistry of starch fraction. III. Retrogradation of
amylase. p. 465.

KOZLITSKY. Magyar Tudományos Akademia. Kémi és Földrajzi Osztály.
Budapest, Hungary. Vol. 11, no. 4, 1956.

Monthly list of East European Accession (EEA) L3, Vol. ~~XXXXXXXXXXXX~~
9, no. 2, Feb. 1960
Uncl.

Country : Hungary D
 Category : Cosmochemistry. Geochemistry. Hydrochemistry.
 Abs. Jour. : Ref Zhur-Khimiya, No 6, 1959 18957
 Author : Ganti, T.
 Institut. :
 Title : Pisolites and Pisolitelike Formations.
 Orig Pub. : Acta mineral.-petrogr. Szeged., 1957, 10,
 15-18

Abstract : Four instances of formation of pisolites are considered: 1) uniform grains of sand covered with water, on rolling in a moving current, are deposited in a form similar to peas; 2) well developed crystals of calcite become coated with a layer of calcite, by crystallization; 3) formation of pisolites takes place in cold water, when the previously separated particles become coated, on condensation, with a thin, stable layer of calcite; 4) aragonite, which separates out from the water during a slow flow of the stream, is converted to the stabler calcite of greater specific gravity. In the course thereof the coating layer of calcite becomes a blistered pisolite-like formation which

Card: 1/2

Country : Hungary D
Category : Cosmochemistry. Geochemistry. Hydrochemistry.
Abs. Jour. : Ref Zhur-Khimiya, No 6, 1959 18957
Author :
Institut. :
Title :

Orig. Pub. :

Abstract : is called "rose stone". -- E. Chepizhnaya.

Card: 2/2

D-7

GANTI, T

HUNGARY/Cosmochemistry, Geochemistry, Hydrochemistry.

D.

Abs Jour : Ref Zhur - Khimiya, No 12, 1958, 39261

Author : Ganti, T

Inst : -

Title : Chemical Conditions for the Formation of Cavities.

Orig Pub : Hidrol. Kozlony, 1957, 37, No 3, 285-288

Abstract : The formation of cavities in limestones cannot be completely explained by the theory of dissolving and water erosion. For, otherwise, they should be also present in dolomites and this is not often the case. However, it is noted that the ratio of calcium to magnesium, which for the common waters is equal to one to one, is sharply changed to 100 to one for the cavities waters. This cannot be explained only by the theory of the secondary solution according to which only calcium carbonate is precipitated from the cavities waters, and the magnesium remains in solution. By saturating and

Card 1/2

HUNGARY/Cosmochemistry, Geochemistry, Hydrochemistry.

D.

Abs Jour : Ref Zhur - Khimiya, No 12, 1958, 39261

supersaturating those waters, calcium carbonate can be precipitated, and the limestone dissolved simultaneously. The volume of the dissolved rocks, calculated on the magnesium content in water is considerably larger than that calculated on the carbonate content. Thus, the effect of the secondary dissolving is considerably higher than that of the primary one. The cavities might be formed only when a secondary dissolving is possible. In dolomites, the dissolving does not occur due to particular conditions of crystallization.

Card 2/2

GANTI, Tibor

The karstic region at Naszaly. Hidrologiai Kozlony 37 no.4:
380-384 '57.

HUNGARY/Physics of Solid Bodies - Structural Crystallography

E-4

Abstr Jour : Ref Zhur - Fizika, No 4, 1959, No 5619

Author : Ganti Tibor

Inst : -

Title : Plastic Capillaries for X-ray Diffraction Investigations

Orig Pub : Magyar fiz. folyoirat, 1958, 6, No 1, 43-46

Abstract : Description of the application of polyvinyl acetate capillaries for powdered specimens (diameter of the capillary is 0.28 mm, thickness of the wall is 0.02 mm) in the taking of Debyeograms.

Card : 1/1

2 7
X-ray diffraction of water-alcohol mixtures. Ganti
Tibor. Magyar Fiz. Folyóirat 7, 10-38 (1959). Water takes
part with unchanged structure up to an EtOH concn. of
80%. Above that the water structure disappears. Water
is consumed completely for hydration, probably one mol.
water for 2 alc. with a H bond between water and alc. This
assocn. structure is stable at 0°.

B. Rona

GANTY, Tibor

Stereochemical structure of proteins. *Magy kem lap* 15 no.4:162-167 Ap '60.

1. Erjedesipari Vallalat.

GANTI, Tibor; NOVAK, Ervin

General paper chromatographic developing process. Magyar
folyoir 68 no.7:293-296 J1 '62.

1. Budapesti Elészto gyar (for Ganti). 2. Orszagos Kozegeszsegugyi
Intezet, Budapest (for Novak).

GANTI, Tibor; J. NAGY, Eva

The effect of x-ray and ultraviolet irradiation on the gemmation of yeast cells treated in latent phase. Magy. radiol. 15 no.4: 239-246 Ag '63.

1. A MTA Orvos-Radiologiai Kutatocsoportjanak (vezeto: Zsebck Zoltan dr.) kozlemenye.

(ULTRAVIOLET RAYS) (SACCHAROMYCES)
(RADIATION EFFECTS)

GANTI, Tibor

Viruses and bacteriophages and their role in the food industry.
Elelm ipar 17 no.12:377-382 D '63.

1. Budapesti Elesztogyar.

FODOR, Jozsef; GANTI, Tibor

Reaction kinetic investigation of the decomposition of diphosphopiridine-nucleotide. Magy kem folyoir 69 no.2:63-66 F '63.

1. Budapesti Elesztogyar, Budapesti.

L 18782-63

EWI(1)/EWI(m)/BDS/ES(j)

AMD/ASD/AFPTC AR/K

ACCESSION NR: AP3005989

H/0021/63/000/004/0239/0246

AUTHOR: Ganti, Tibor; Nagy, Eva J.

TITLE: Effect of X- and ultraviolet radiation on the gemmation of yeast cells treated in the latent phase

SOURCE: Magyar radiologia, no. 4, 1963, 239-246

TOPIC TAGS: gemmation, yeast cell., Saccharomyces cerevisiae, synchronism, lag phase, ultraviolet irradiation, X-irradiation, cellular radiation resistance

ABSTRACT: The authors investigated the changes brought about in the first gemmation of Saccharomyces cerevisiae by UV- and X-irradiation. It was found that UV decreases the degree of synchronism and the number of cells capable of gemmation; however, when the irradiation is carried out during the late part of the "lag" phase, this inhibition is not yet observable at the beginning of gemmation. The cause of this phenomenon may be the increased resistance of the cells to radiation immediately prior to gemmation, or that the UV effect requires a certain length of time to develop. In the case of X-irradiation gemmation set in about 10 minutes later than in the case of unirradiated cells when

Card 1/2

L 18782-63

ACCESSION NR: AP3005989

the irradiation was carried out during the early period of the "lag" phase; when the irradiation was effected immediately prior to the start of germination, the delaying effect was observable only after a certain time. Orig. art. has: 8 figures.

ASSOCIATION: MTA orvos-radiologiai kutatocsoport (MTA Medical-Radiological Research Group)

SUBMITTED: 00

DATE ACQ: 27Aug63

ENCL: 00

SUB CODE: AM

NO REF SOV: 000

OTHER: 016

Card 2/2

ACCESSION NR: AP4032719

H/0021/64/000/002/0110/0117

AUTHOR: Ganti, Tibor (Doctor)

TITLE: Use of synchronous cell-cultures in radiological examinations

SOURCE: Magyar radiologia, no. 2, 1964, 110-117

TOPIC TAGS: radiology, radiation sensitivity, cell irradiation, cell culture, synchronous culture, cell multiplication, mitosis, morphological analysis, cell morphology, biochemical analysis, cell chemistry

ABSTRACT: Cell-cultures in which the majority of the cells at any moment is in the same phase of the reproductive cycle (synchronous cultures), offer a possibility of examining the specific processes of cell-multiplication and their sensitivity to radiation. A great number of physical and chemical methods has been worked out for the synchronization of cell-cultures. The effect of the radiation on the mitosis and the radiation sensitivity in its various stages may be examined in synchronous cultures by means of morphological and biochemical methods. Orig. art. has: 2 figures.

Cord 1/2

ACCESSION NR: AP4032719

ASSOCIATION: none

SUBMITTED 00

DATE ACQ: 07May64

ENCL: 00

SUB CODE: LS

NO REF SOV: 000

OTHER: 065

Card 2/2

GANTI, T.

Fermentation by synchronously dividing yeast cells in nitrogen-poor medium. Acta microbiol. acad. sci. Hung. 11 no.2:147-153 '64.

1. Yeast Factory, Budapest.

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Studies on the kinetics of NAD-decomposition. Acta physico-
chem. sci. Hung. 26 no.3:199-205 '6

1. Yeast Factor, Budapest.

L 43013-66

ACC NR: AT6031821

SOURCE CODE: HU/2505/65/026/003/0199/0205
42
5+1

AUTHOR: Ganti, Tibor; Fodor, Jozsef--Fodor, Y.

ORG: Yeast Factory, Budapest; I. Surgical Clinic, Medical University of Budapest, Budapest (Budapesti Orvostudományi Egyetem, I. Sebészeti Klinika)

TITLE: Studies on the kinetics of NAD decomposition

SOURCE: Academia scientiarum hungaricae. Acta physiologica, v. 26, no. 3, 1965, 199-205

TOPIC TAGS: yeast, chemical kinetics, aqueous solution, activation energy, plant chemistry

ABSTRACT: An attempt has been made to establish the optimal conditions of nicotine amide adenine dinucleotide (NAD) extraction from yeast, taking into account the kinetics of extraction and the decomposition of the compound. The results obtained supported the findings of LOWRY et al. (1961), concerning the rate of decomposition of NAD. The Arrhenius diagram, plotted on the basis of computed k values (decomposition rate constant of NAD) is a straight line. The temperature gradient, activation energy and frequency factor calculated from the measured data were within the expected range. The decomposition rate constant of NAD in an aqueous solution showed a deviation from the rate in the extracted yeast. The characteristic constants were: temperature gradient $2.5/10^\circ \text{C}$; activation energy 26.98 kcal/mole ; frequency factor 0.59×10^{16} ; optimal temperature 80°C ; period of extraction 5 minutes. Orig. art. has: 3 figures, 3 tables and 3 formulas. [Orig. art. in Eng.] [JPRS]

SUB CODE: 06, 07 / SUBM DATE: 18May63 / OTH REF: 006

Card 1/1 MLP
0914 0369

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LIST AND INDEX GROUPS																									
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<p>15</p> <p>—Oxidation-reduction status and the principal characteristics of the soil. J. I. Gantimayr, Z. Panteonovsk, Dzungu Rodenk. <i>IS. AC. AS (USSR)</i>. The results of exper. with composts of soil and sewage sludge are discussed, and phenomena of sewage sick and waterlogged soils explained on the basis of oxidation-reduction processes taking place therein. An electrometric method for detg. oxidation-reduction status is described. Soil equiv. to about 10 g. dry is rubbed up with 10-500 mg. quinhydrone and 0.8 g. Mg biphthalate for buffer. A Pt electrode about 1 cm. square is buried in the mass, and after 10 mins. the potential vs. a standard half-cell is detd. in the usual manner. The difference between the e. m. f. in the soil-quinhydrone-buffer mixt. and that in a quinhydrone-buffer mixt. is a measure of the oxidizing or reducing power of the soil, and is expressed as mv. + or -. It is concluded that the</p> <p>(over)</p>																									
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moisture content and permeability to air det. the nature of the biol. processes which take place in soil with large amts. of decaying org. matter. The waterlogging or gelling processes in sewage sprinkling or filtration fields are characterized: pH is raised to 7 or more; the value of E_h is decreased and also the ability to change e. m. f. in the quinhydrone-buffer treatment described; there is a mobilization and leaching out of Fe and great changes in the inorg. colloids; NH_4 becomes an important exchangeable base. It is stated that E_h , the electrode potential, and r_h , the index of oxidation-reduction status calcd. therefrom, are detd. by the method of Remesov (cf. C. A. 28, 28234); the procedure is described in detail, but the necessary calens. are not explained. Diminished values for E_h and r_h indicate that conditions are less favorable for crop growth. Thirty-one references. C. I. S.

PROCESS AND PROPERTIES INDEX																									
<p><i>CH</i></p> <p>The method for the determination of ferrous and ferric iron in soils irrigated with rain water. I. I. Cantimurov. <i>J. Applied Chem. (U. S. S. R.)</i> 10, 1504-15 (in German) 1513 (1937).—Extn. of soil with 1.0 N H₂SO₄ did not reduce ferric to ferrous ion. The acid ext. can be titrated for ferric and ferrous ions by the Knecht method or iodometrically. Twelve references. A. A. Podgorny</p>																									
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<p>GROUPS</p>																									

[illegible]

Determination of ability of soils to change the potential of the quinone-hydroquinone system. L. I. Gaitanovskaya: *J. Applied Chem.* (U. S. S. R.), 12, 415-20 (in French, 420) (1939).—The samples of soil (the moisture content not less than 30%) were mixed with buffer solutions of pH 4.2 (Mg biphthalate and phthalic acid), 5.2 (K biphthalate and NaOH), and 6.4 (borax, H_2SO_4 and NaCl). Each series of samples (pH 4.2, 5.2 and 6.4) was mixed with hydroquinone, the amt. of which increased in each series from 0 to 700 mg. per 15 g. of soil. Potentials are tabulated. A. A. Podgorny

A. A. Podgornyy

A S M - I L A METALLURGICAL LITERATURE CLASSIFICATION

PROCESSAL AND PROPERTIES																									
<div style="display: flex; justify-content: space-between;"> <div> <p>CA</p> </div> <div> <p>15</p> </div> </div>																									
<p>K. K. Gedroiz and the science of the soil. I. I. Ganti... ustrov. Pedology (U.S.S.R.) 1947, 207-12; Chem. Zentr. 1947, 1, 700; cf. preceding abstr. M. G. Moore</p>																									
<div style="display: flex; justify-content: space-between;"> <div> <p>ASB-SLA DETALLURGICAL LITERATURE CLASSIFICATION</p> </div> <div> <p>STORI 1116314</p> </div> </div>																									
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GANTIMUROV, I. I.

"Research on the Problems of Sanitary Soil Science." Sub 12 Jan 51,
Moscow Order of Lenin State U imeni M. V. Lomonosov.

Dissertations presented for science and engineering degrees in
Moscow during 1951.

SO: Sum. No. 480, 9 May 55

CA

15

Electrodialysis of soils as an oxidation-reduction process.
 I. I. Gantimurov (Agr. Inst., Omsk) *Kolloid Zh.* 13, 94-8 (1951).--Electrodialysis of soils does not consist in the removal of ions only; the oxidation-reduction potential E also changes. Thus, E of a soil having pH 7.3 and elec. cond. of an ext. of less than $0.008 \text{ ohm}^{-1} \text{ cm}^{-1}$ was lowered from +0.35 to -0.30 v. near the cathodic membrane after 4 hrs. of electrodialysis; later E rose to +0.30 v. The potential near the anodic membrane was but little changed by electrodialysis. The mud of a salt lake gave an ext. of pH 8.2 and elec. cond. $0.08 \text{ ohm}^{-1} \text{ cm}^{-1}$; its E was -0.5 v. and after 33 hrs. of electrodialysis rose to 0.45 v. near the anode and 0.0 v. near the cathode; the pH became 4.3 at the anode and 10.2 at the cathode. The soil near the cathodic membrane is visibly different from that near the anodic. Apparently, these two layers differ in the degree of oxidation of their org. substances, Fe and Mn, and in some other properties.
 J. J. Bikerman

GANTIMUROV, I.I. (Omak)

The solonchik and bog processes in the watershed of the river Om.
Pochvovedenie '52, 1007-11. [No. 11] (MLRA 5:12)
(CA 47 no.13:6590 '53)

ORIGIN : USSR
SUBJECT : Soil Science. Tillage. Improvement. Erosion. J
JOUR. NAME : RZhBiol., No. 3 1959, no. 10727
AUTHOR : Gantimurov, I. I., Koshkin, N. A.
INSTITUTION : Ubinskaya Experiment and Soil Improvement Station
TITLE : Measures for the Reclamation and Improvement of
Low-Fertility Lands in the Regions of Baraba.
ORIG. PUB. : Izv. nauchno-issled. i opyt. malior. st., 1957, No. 2,
5-12
ABSTRACT : No abstract.

CARD: 1/1

49

USSR / Soil Science. Cultivation. Melioration, Erosion. J

Abs Jour: Ref Zhur-Biol., No 21, 1958, 95775.

Author : Gantimurov, I. I.

Inst : Ubinskoye Experiment Melioration Station.

Title : Results of an Experiment with Methods of Primary Cultivation of Marshes.

Orig Pub: Byul. nauchno-issled. i opytn. rabot Ubinsk. opytn. melior. st., 1957, No 2, 30-41.

Abstract: Results are described of experiments conducted on a lowland sedge-reed weakly-saline marsh in the Ubinskoye Experimental-Improvement Station (Baraba) in 1955-1956. Primary cultivation was conducted at the end of the summer in 1955 without allowing the turned-up virgin soil to lie fallow. Various variants of cultivation were used (cutting, deep plowing in different combina-

Card 1/2

USSR / Soil Science. Cultivation. Melioration, Erosion. J

Abs Jour: Ref Zhur-Biol., No 21, 1958, 95775.

Abstract: tions). In the spring of the following year, the cultivated plots were rolled. During sowing of oats, 2 c per 1 ha of P_c and 1 c per 1 ha of N_{aa} were applied. Data are cited of phenological observations for the development of the plants and harvest capacity of some soil-agrochemical indicators. The highest harvest with fertilizers and without fertilizers was obtained by plowing with a shrub-marsh plow plus disking plus cutting. -- F. N. Sofiyeva.

Card 2/2

USSR/Soil Science - Genesis and Geography of Soils.

J

Abs Jour : Ref Zhur Biol., No 22, 1958, 99989

Author : Gantimurov, I.I.

Inst : Novosibirsk Agricultural Institute

Title : Soil Conditions in the Basic Wood Types of the Former
"Dorovoye" National Forest Kazakh SSR

Orig Pub : Tr. Novosib. s.-kh. in-t, 1957, 12, 92 pages, ill.

Abstract : Under the pine trees of the "Dorovoye" National Forest
soils of the podzol type are being formed on granites,
which are characterized by a coarse mechanical structure
and by a comparatively small depth of the soil profile.
The soil morphology and peculiarities of its distribution
are examined in detail. A study of the soils' chemical
properties indicated that the free Al predominates
over the exchangeable H ion. The exchangeable

Card 1/2

USSR/Soil Science - Genesis and Geography of Soils.

J

Abs Jour : Ref Zhur Biol., No 22, 1958, 99989

acidity of the lower horizons is caused only by the free
Al and appears to be a characteristic of the subsoils
conditioned by the weathering processes of soil-forma-
tion rocks - granite, quartz and mica. The character of
forest-cultivating conditions and the special features
of the soils' water regime are presented. -- P.V.
Shranko

Card 2/2

- 20 -

GANTIMUROV, I.I.

Soils conditions of Omsk and Novosibirak in relation to the
landscape gardening. Trudy TSSBS no.3:51-56 '60. (MIRA 15:3)
(Omsk--Landscape gardening)
(Novosibirsk--Landscape gardening)

GANTIMUROV, I.I.; POGODAYEVA, I.S.

Upland and crustal Solonetz soils in the districts of Baraba
and methods for their reclamation. Trudy Biol. inst. Sib. otd.
AN SSSR no. 242-246 '62 (MIRA 17:8)

GANTIMUROV, I.I.; BASHIROVA, F.N.; TARASOVA, H.V., red.

[Scientific bases for studying soil conditions in cities] Nauchnye osnovy izucheniia pochvennykh uslovii v gorodakh. Novosibirsk, Red.-izd. otdel Sibirskogo otd-niia AN SSSR, 1964. 135 p. (MIRA 18:1)

GANTIMUROV, P.G., inzh.; VISHNEVSKIY, N.I.; RYSIN, V.I., inzh.;
BANDIN, M.M.

Exchange of practices by the enterprises of economic councils.
Torf. prom. 39 no.5:29-33 '62. (MIRA 16:8)

1. Sverdlovskiy sovet narodnogo khozyaystva (for Gantimurov).
2. Glavnyy energetik torfopredpriyatiya "Krasnoye znanya"
Belorusskogo soveta narodnogo khozyaystva (for Vishnevskiy).
3. Torfopredpriyatiye Radovitskiy Mokh Moskovskogo oblastnogo
soveta narodnogo khozyaystva (for Rysin). 4. Leningradskiy
gosudarstvennyy trest torfyanoy promyshlennosti (for Bandin).

GANTIMUROVA, N.

COUNTRY : USSR
CATEGORY :

M-6

ABS. JOUR. : RZBiol., No. 1/2, 1958, No. 8/119

AUTHOR : GANTIMUROVA, N.
INST. : Shirokaya land reclamation experiment station
TITLE : Sensitivity of Corn to the Degree of Soil Salination

ORIG. PUB. : Ryul. zhurnal-izvest. i opyt. rabot Uzbek. opyt. melior. st., 1958, No 2, (8-6)

ABSTRACT : At the station, analyses were made of soil samples from different parts of a field planted with corn, the samples being collected from those parts where a difference in the condition of the plants was apparent. The analyses showed that yields of green crop of corn depend on pH of the soil and total alkalinity of the soil extract: highest yield was obtained in a soil of pH 6.0, of low electric conductivity and alkalinity, but with a large content of sulfates; the lowest -- in a soil with clear indication of salination, where the pH was 3.60, the aqueous extract showed high electric conductivity, was low in sulfates and had a high total alkalinity.--A.A.Orchibrya

CARD://

CA

3

The Raman effect of oxonium compounds. A. R. Gantmacher, M. V. Vol'kenshtein and Ya. K. Sytkin. *Acta Physicochimica U. R. S. S. 12, 780-82(1946)* (in English); cf. C. A. 33, 7004^g; 34, 4001^g.—From vapour data on the Raman spectra for the system $(CH_3)_2O + n HCl$ with $n = 1, 2, 3$ at temps. from -80 to $+20^\circ C$, V. and S. find that interaction causes a lowering of the valency frequency of the C-O-C system, an increase of that of CH_3 , and of the deformation frequency of C-O-C. For 1:1 molar, the displacement of the CH_3 and C-O-C frequencies is const. throughout the temp. range studied. With increasing amts. of HCl, a new line $\Delta\nu = 600\text{ cm}^{-1}$ and a diffuse band at 2400 cm^{-1} due to HCl deformed 14% relative to liquid HCl, appear simultaneously. The 2400 band is attributed to the oxonium resonance complex $CH_3-O^+ \cdots H-Cl^- \cdots HCl$ and the 500 line to a hindered HCl rotation. The evidence indicates trivalent rather than quadrivalent oxygen; the latter is also unlikely from thermodynamical considerations. F. H. R.

Phys. Chem. Inst. in Karpov, Moscow

ASB-51A METALLURGICAL LITERATURE CLASSIFICATION

1ST AND 2ND ORDERS										PROCESSES AND PROPERTIES UNDER										195 AND 196									
<div style="position: absolute; top: 10%; left: 10%; font-size: 2em; font-family: cursive;">BC</div> <div style="position: absolute; top: 10%; right: 10%; font-size: 2em; font-family: cursive;">A-1</div>										<p style="text-align: center;">Raman effect of certain compounds. A. Gantmakher, M. Volkovskii, and J. Sirkin (<i>Acta Physicochim. U.S.S.R.</i>, 1941, 16, 187--190).--Raman spectra of mixtures of Me_2O with HCl in the mol. ratios 1:1, 1:2, and 1:3 were studied at 20°, -20°, -40°, and -60°. In the 1:1 mixture there is a shift in the Me_2O frequencies, and the frequency due to HCl is absent. In the 1:2 and 1:3 mixtures there is a still greater shift in the Me_2O frequencies, whilst new frequencies appear at 600 cm^{-1} and 2600 cm^{-1}, the latter being the center of a broad diffuse band. The broad band is attributed to strongly deformed HCl, whilst the frequency 600 cm^{-1} is interpreted in the same way as the corresponding internal frequency of liquid H_2O. The data of Briggs and Lauppe (<i>J. A.</i>, 1937, 1, 399) are analyzed and their interpretation criticized. Raman spectra of solutions of ethers and alcohols in excess of H_2 yields are consistent with the formation of H bonds and afford no satisfactory evidence of a structure with H^+.</p> <div style="text-align: right; margin-top: 10px;">V. L. U'</div>										<div style="font-size: 1.5em; font-family: cursive; text-align: center;">Karpov Phys. Chem. Inst., Moscow</div>									
ASS. SLA METALLURGICAL LITERATURE CLASSIFICATION										1950 1951 1952 1953 1954 1955 1956 1957 1958 1959 1960 1961 1962 1963 1964 1965 1966 1967 1968 1969 1970 1971 1972 1973 1974 1975 1976 1977 1978 1979 1980 1981 1982 1983 1984 1985 1986 1987 1988 1989 1990 1991 1992 1993 1994 1995 1996 1997 1998 1999 2000 2001 2002 2003 2004 2005 2006 2007 2008 2009 2010 2011 2012 2013 2014 2015 2016 2017 2018 2019 2020 2021 2022 2023 2024 2025 2026 2027 2028 2029 2030 2031 2032 2033 2034 2035 2036 2037 2038 2039 2040 2041 2042 2043 2044 2045 2046 2047 2048 2049 2050 2051 2052 2053 2054 2055 2056 2057 2058 2059 2060 2061 2062 2063 2064 2065 2066 2067 2068 2069 2070 2071 2072 2073 2074 2075 2076 2077 2078 2079 2080 2081 2082 2083 2084 2085 2086 2087 2088 2089 2090 2091 2092 2093 2094 2095 2096 2097 2098 2099 2100 2101 2102 2103 2104 2105 2106 2107 2108 2109 2110 2111 2112 2113 2114 2115 2116 2117 2118 2119 2120 2121 2122 2123 2124 2125 2126 2127 2128 2129 2130 2131 2132 2133 2134 2135 2136 2137 2138 2139 2140 2141 2142 2143 2144 2145 2146 2147 2148 2149 2150 2151 2152 2153 2154 2155 2156 2157 2158 2159 2160 2161 2162 2163 2164 2165 2166 2167 2168 2169 2170 2171 2172 2173 2174 2175 2176 2177 2178 2179 2180 2181 2182 2183 2184 2185 2186 2187 2188 2189 2190 2191 2192 2193 2194 2195 2196 2197 2198 2199 2200 2201 2202 2203 2204 2205 2206 2207 2208 2209 2210 2211 2212 2213 2214 2215 2216 2217 2218 2219 2220 2221 2222 2223 2224 2225 2226 2227 2228 2229 2230 2231 2232 2233 2234 2235 2236 2237 2238 2239 2240 2241 2242 2243 2244 2245 2246 2247 2248 2249 2250 2251 2252 2253 2254 2255 2256 2257 2258 2259 2260 2261 2262 2263 2264 2265 2266 2267 2268 2269 2270 2271 2272 2273 2274 2275 2276 2277 2278 2279 2280 2281 2282 2283 2284 2285 2286 2287 2288 2289 2290 2291 2292 2293 2294 2295 2296 2297 2298 2299 2300 2301 2302 2303 2304 2305 2306 2307 2308 2309 2310 2311 2312 2313 2314 2315 2316 2317 2318 2319 2320 2321 2322 2323 2324 2325 2326 2327 2328 2329 2330 2331 2332 2333 2334 2335 2336 2337 2338 2339 2340 2341 2342 2343 2344 2345 2346 2347 2348 2349 2350 2351 2352 2353 2354 2355 2356 2357 2358 2359 2360 2361 2362 2363 2364 2365 2366 2367 2368 2369 2370 2371 2372 2373 2374 2375 2376 2377 2378 2379 2380 2381 2382 2383 2384 2385 2386 2387 2388 2389 2390 2391 2392 2393 2394 2395 2396 2397 2398 2399 2400 2401 2402 2403 2404 2405 2406 2407 2408 2409 2410 2411 2412 2413 2414 2415 2416 2417 2418 2419 2420 2421 2422 2423 2424 2425 2426 2427 2428 2429 2430 2431 2432 2433 2434 2435 2436 2437 2438 2439 2440 2441 2442 2443 2444 2445 2446 2447 2448 2449 2450 2451 2452 2453 2454 2455 2456 2457 2458 2459 2460 2461 2462 2463 2464 2465 2466 2467 2468 2469 2470 2471 2472 2473 2474 2475 2476 2477 2478 2479 2480 2481 2482 2483 2484 2485 2486 2487 2488 2489 2490 2491 2492 2493 2494 2495 2496 2497 2498 2499 2500 2501 2502 2503 2504 2505 2506 2507 2508 2509 2510 2511 2512 2513 2514 2515 2516 2517 2518 2519 2520 2521 2522 2523 2524 2525 2526 2527 2528 2529 2530 2531 2532 2533 2534 2535 2536 2537 2538 2539 2540 2541 2542 2543 2544 2545 2546 2547 2548 2549 2550 2551 2552 2553 2554 2555 2556 2557 2558 2559 2560 2561 2562 2563 2564 2565 2566 2567 2568 2569 2570 2571 2572 2573 2574 2575 2576 2577 2578 2579 2580 2581 2582 2583 2584 2585 2586 2587 2588 2589 2590 2591																			

Change in Raman spectra of chloroprene and isoprene in the polymerization process. A. Ginzman and S. Madvedev *Izda. Physicochim. U.R.S.S.*, 1942, 14, 1-11).—Raman data are given for chloroprene (I), a 30% solution of polymerised (II) in the monomer, a solution of the polymer in C_6H_6 or CCl_4 , and the pure polymer; corresponding data are also given for isoprene (III). In each case polymerisation gives rise to a non-conjugate double-bond line (1660 cm^{-1} in (I) and 1665 cm^{-1} in (II)) not given by the monomer; a conjugate double-bond line ($1630, 1640\text{ cm}^{-1}$) decreases in intensity as polymerisation proceeds, but remains much stronger than the $1660-1665\text{ cm}^{-1}$ line when >50% of polymer is present. The 1020 cm^{-1} C-H line given by (I) is displaced to 1003 cm^{-1} on polymerisation, and its intensity increases considerably. Changes in C-C and C-Cl lines are discussed. Polymerisation does not increase the background intensity of the spectrograms. A. J. E. W.

Kayser Inst. Phys. Chem., Moscow

GANTMAKHER, A. R.; NEPOMNICHENKO, S. S.

Physico-Chemical Institute imeni L. Ya. Karpov, Moscow (-1941-)

"The Change of the Raman Spectrum of Chloroprene and Isoprene in the Polymerization Process." Zhur. Fiz. Khim. Vol 17, No 1, 1943.

BR-52059019

Catalytic polymerization of unsaturated compounds.
Catalytic polymerization of styrene. A. R. Gantmakher
and S. S. Medvedev. *Izvestiya Akad. Nauk SSSR Khim. Soedinenii, Doklady 6-oi Konf. Vysokomolekul. Soedinenii*
yuzn, Akad. Nauk S.S.S.R. 1949, 50-53; cf. C.A. 43,
7205f; 47, 8480c. — Dilatometric and viscometric study of
polymerization of PhCH=CH₂ was made at 0°, 10°, 20°, and
30° in EtCl soln. with SnCl₄ as catalyst. The rate is well
defined by the equation: $-d[I]/dt = K_1 K_2 [B_0] [A] / (K_1 + K_2 [A])$, where B_0 is initial catalyst concn., K_1 is the
rate const. of initiation of chains (about 2.102×10^{-3} mole/l.
min.), K_2 is chain growth rate const. (1.6×10^{-4} mole/l.
min.), K_3 is chain rupture rate const. (1.035×10^{-3} min.⁻¹).
This is derived on the assumption that catalyst units remain
bound to the growing chain. In this process the recombina-
tion mechanism of chain rupture is absent. Characteristic
viscosity is proportional to the square root of initial concn.
of monomer and changes from 2.75 at 0° to 1.8 at 30°. HCl
and H₂O accelerate the reaction; the former also accelerates
chain rupture. At high concn. of monomer there was ob-
served an acceleration of the polymerization rate as the pro-
cess proceeded; this is attributed to branched chain mecha-
nism and is absent in peroxide-catalyzed reactions.

G. M. Kosolapoff

(1) AND (20) COPIES		PROCESSES AND PROPERTIES INDEX		(10) AND (20) COPIES	
<div style="font-size: 2em; font-weight: bold; margin-bottom: 10px;">CA</div> <div style="font-size: 0.8em;">COMMON SUBJECTS</div> <div style="font-size: 0.8em;">COMMON MATERIALS INDEX</div>		<p>Catalytic polymerization of unsaturated compounds A. R. Gantmakher and N. N. Malyshev <i>Zhne. Fiz. Khim.</i> 23, 815-20 (1940). Polymerization of styrene (I) at 20° in the presence of SnCl₄ in EtCl was detd. in a dilatometer. If <i>A</i> was the variable concn. of monomer I and <i>B</i> the const. concn. of SnCl₄, $-dA/dt = k_1 B A^2 / (k_2 + k_1 A)$, when the starting <i>A</i> (<i>A</i>₀) varied between 0.4 and 1.4 mol./l. and <i>B</i> was 0.00-0.06 mol./l. The const. <i>k</i>₁ of the formation of the first active center was 0.0021./mol. min. and <i>k</i>₂ of the rate of chain breaking was 0.001 min.⁻¹. <i>k</i> is the const. of formation of the complex of I and SnCl₄. The equation shows that addn. of this complex to another mol. of I is the rate-detg. process. The temp. coeff. of the reaction was 1.37 between 0 and 30°. When <i>A</i>₀ was 2.5 <i>M</i>, $(-dA/dt)$ increased "auto-catalytically" showing branching of reaction chains; this effect was absent in the presence of SnCl₄ + HCl or of (BuO)₂. From the viscosity of polymer solns. in CCl₄ at 20°, $[\eta] = \lim (\eta_{sp}/c)$ was calcd. It was independent of <i>B</i>, almost independent of the degree of polymerization, and was 2.1 √<i>A</i>₀ for polymers produced at 20°; after polymerization at 0° $[\eta]$ was 1.6 times that after polymerization at 30°. As $[\eta]$ was independent of <i>B</i>, chains are not broken by recombination. HCl added to SnCl₄ accelerated the polymerization more the more HCl was added until the mol. ratio 1:1 was reached. The $[\eta]$ of the polymer obtained in the presence of SnCl₄ + HCl was about 1/2 that in the presence of SnCl₄. H₂O acts similarly to HCl. Presumably H ions are the active component of HCl.</p> <p style="text-align: right;">J. J. Dikerman</p>			
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GANTMAKHER, A. R.

PA 196717

USSR/Chemistry - Plastics

Nov 51

"Catalytic Polymerization of Unsaturated Compounds.

II. Effect of the Dielectric Constant of the Me-

diu on the Catalytic Polymerization of Styrene,"

A. R. Gantmakher, S. S. Medvedev, Phys. Chem Inst

Imeni L. Ya. Karpov, Moscow

"Zhur Fiz Khim" Vol XXV, No 11, pp 1328-1334

Ionic character of processes of styrene polymeri-
zation with SnCl_4 was shown by investigation of
processes in solvents with different dielec consts
(i.e., higher dielec const increases reaction rate
HCl admixts have different effects in different

196717

USSR/Chemistry - Plastics (Contd)

Nov 51

solvents, accelerating reaction in ethylchloride,
dichloroethane, and o-nitrotoluene, lowering re-
action rate in cyclohexane, and lowering mol wt
when solvent has different dielec const than HCl)
Found that reaction rate is proportional to
square of concn of styrene in ethyl chloride
with uniform dielec const. Proposes eq for de-
pendence.

196717

GANTMAKHER, A. R.

USSR/Chemistry - Elastomers; Plastics Feb 52

"Catalytic Polymerization of Unsaturated Hydrocarbons.
III. Catalytic Polymerization of Isoprene," A. R.
Gantmakher, B. S. Medvedev, Phys Chem Inst Issled
L. Ya. Karpov, Moscow

"Zhur Fiz Khim" Vol XXVI, No 2, pp 173-179

Results on the catalytic polymerization of isoprene
with SnCl_4 in an EtCl soln agree with those obtained
on styrene. The rate of reaction increases with
time, because there is an autocatalytic increase of
the number of active centers in the branched chain
reaction. Branching of reaction chains occurs under

211745

participation of SnCl_4 . In agreement with the
proposed mechanism, EtCl inhibits acceleration of
reaction and lowers mol wt of products. At -80° ,
the branched chain polymerization of isoprene does
not occur even with a 100% monomer. Interaction of
growing chains with double bonds of polymer mols
proceeds under reduction of deg of unsatn. Reduction
of the concn of monomer results in decrease of deg
of unsatn and increase of the temp of gelation.

211745

GANTMAKHER, A. P.

USSR/Chemistry - Elastomers, Polymerization 21 Oct 52

"Initiation Mechanism in the Catalytic Polymerization of Unsaturated Compounds," A. P. Gantmakher, S. S. Medvedev, Corr Mem, Acad Sci USSR, T. E. Lipatova

"Dok Ak Nauk SSSR" Vol 86, No 6, pp 1109-1111

A conclusion drawn by P. H. Fleisch, M. Polanyi, H. A. Skinner, A. S. Evans, and G. W. Meadows is found to be incorrect. These authors claim that in the catalytic polymerization of isobutene in the liquid state or in hexane solus with $TiCl_4$ and BF_3 catalysts at

234T25

low temps, the addn of H_2O , CCl_3COOH , etc., is necessary in order that the reaction may proceed. This is disputed in the present work: It is shown that by raising either the temp or the dielec const of the solvent, catalytic polymerization of unsatd compds can be made to proceed in the presence of aprotonic acids ($SnCl_4$, $AlCl_3$, BF_3 , $TiCl_4$, and others) without the addn of H_2O , CCl_3COOH or other substances acting on the catalyst under formation of protonic acids.

234T25

GANTMAKHER, A.R.

USSR/ Catalytic polymerization

Card 1/2 Pub. - 21/52

Authors : Gantmakher, A. R., and Medvedev, S. S. Act. Memb. of Acad. of Sc. USSR

Title : ~~Investigation of the kinetics of combined catalytic polymerization~~
Some peculiarities of the kinetics of combined catalytic polymerization

Periodical : Dok. AN SSSR, 100/2, 275-278, Jan 11, 1955

Abstract : Investigation was made to determine the kinetics of combined and separate catalytic polymerization of isoprene-butadiene and styrene-butadiene systems. The polymerization was carried out in EtCl with SnCl₄ as catalyst at 0° by the dilatometric method. Results indicate that the butadiene polymerization does not take place at temperatures of from 0 to 20°; the polymerization with this less active catalyst requires much higher temperatures. Isoprene with SnCl₄ was observed to polymerize at temperatures ranging from 0 to -80°.

Institution : The L. Y. Karpov Scientific Research Phys-Chem. Institute

Submitted : July 3, 1954

Periodical : Dok. AN SSSR, 100/2, 275-278, Jan 11, 1955

Card 2/2 Pub. 22 - 21/52

Abstract : Initial and maximum rate of polymerization were increasing after adding a relatively large amount of butadiene to the isoprene. An entirely different effect was observed during the addition of butadiene to styrene. Eight references: 2 English; 2 USA; 1 German and 3 USSR (1923-1951). Table; graph.

CANTMAKHER, A.R.

USSR/Chemistry - Catalytic polymerization

Card 1/1 Pub. 22 - 22/49

Authors : Lipatova, T. E.; Cantmakher, A. R.; and Medvedev, S. S. Memb. Corresp.
of Acad. of Sc., USSR

Title : Catalytic copolymerization of unsaturated compounds

Periodical : Dok. AN SSSR 100/5, 925-928, Feb 11, 1955

Abstract : The kinetics of catalytic copolymerization of isoprene-styrene, isoprene-alpha-methylstyrene systems over SnCl_4 catalysts was investigated to determine the copolymerization constants for these systems. The dependence of the copolymerization rate upon the composition of the basic mixture and the molecular weights of the homologous copolymers were determined. The ion mechanism of catalytic polymerization in the presence of an SnCl_4 catalyst is explained. Eight references: 6 USA and 2 USSR (1944-1955). Table; graphs.

Institution :

Submitted : July 3, 1954

GANTMAKHER, A. R.

✓ The simultaneous carbonium polymerization of unsaturated compounds. Simultaneous polymerization of isoprene with styrene. T. E. Lindtsov, A. R. Gantmakher, and S. S. Medvedev (L. Ya. Kerner Phys. Chem. Inst., Moscow). Zhur. Fiz. Khim. 30, 1752-4 (1956).—The rate of polymerization was studied in relation to the compn. of the starting mixt. The mol. wt. and the compn. of the polymers were detd. The data show that 3 components participate in the polymerization: isoprene, styrene, and the copolymer. The data on the compn. of the copolymer show that for a high concn. of styrene in the starting mixt., the isoprene is the more reactive monomer and the final copolymer is enriched in its isoprene content. L. Bortas-Lenci.

Meals

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GANTMAKHVER, A.R.

Mechanism of catalytic polymerization of unsaturated compounds
 A. R. Gantmakher and S. S. Medvedev (*Dokl. Akad. Nauk SSSR*
 1986, 198, 1031-1034). Polymerization of vinyl alkyl ethers in
 presence of I_2 consists of the reactions: $2M_1 \rightarrow M_1I + I_2$
 (initiation of chain); $M_1I + M \rightarrow M_2I$ (growth of chain); $M_2I \rightarrow$
 $M_1I + M$ and $M_2I + I_2 \rightarrow M_1I + M_2I$ (breaking of reaction
 chains), for which the velocity const. are k_1, k_2, k_3 and k_4 respectively.
 If $k_2 \ll k_3$, the velocity of polymerization is given by:
 $k_1 k_2 [I_2]^2 [M] / k_4$, where M represents monomer. At high $[M]$, $k_2 \ll k_3$
 and $v \approx k_1 k_2 [I_2]^2 [M] / (k_3 + k_4 [M])$.
 R. I. Iosadze.

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CAITMANTLER, A. R. and MEDVEDEV, S. S.

"Kinetics of copolymerization," a paper presented at the 9th Congress
on the Chemistry and Physics of High Polymers, 28 Jan-2 Feb 57, Moscow,
Karpov, Inst.

B-3,084,395

SOV/76-32-9-13/46

AUTHORS: Lipatova, T. E., Gantmakher, A. R.,
Medvedev, S. S.

TITLE: The Catalytic Copolymerization of Unsaturated Compounds
(Sovmestnaya kataliticheskaya polimerizatsiya nenasyszhennykh
soyedineniy) II. The Copolymerization of Isoprene and α -
Methyl Styrene (II. Sovmestnaya polimerizatsiya izoprena s
 α -metilstirolom)

PERIODICAL: Zhurnal fizicheskoy khimii, 1958, Vol 32, Nr 9,
pp 2029 - 2034 (USSR)

ABSTRACT: These compounds were polymerized in an ethyl chloride
solution at 0°C using tin (IV) chloride as a catalyst. The
composition of the copolymers formed was determined from
the volume decrease during the polymerization and by means
of infra-red spectroscopy. The spectra are reproduced in
figures 1, 2, and 3. The molecular weights were determined
by the osmotic method (Table 2). Diagrams show the course
of the polymerization of α -methyl styrene alone (Fig 5) and
of the copolymerization of isoprene and α -methyl styrene.
The work shows that three components, isoprene, α -methyl
styrene, and a product of copolymerization with inner

Card 1/2

The Catalytic Copolymerization of Unsaturated Compounds. SOV/76-32-9-13/46
II. The Copolymerization of Isoprene and α -Methyl Styrene

double bonds take part in the polymerization reactions. Steric factors are important in the reaction between the tertiary carbonium ion and monomers. α -methyl styrene is considerably more reactive than isoprene. The initial reaction rate in the polymerization of the α -methyl styrene is decreased by the introduction of isoprene into the system. This is explained by the fact that various complex monomer-catalysts are formed. This formation reduces the concentration of the complex formed by the tin (IV) chloride with α -methyl styrene, which is the more active of the two monomers in initiating carbonium polymerization. There are 7 figures, 2 tables, and 2 references, 2 of which are Soviet.

ASSOCIATION: Fiziko-khimicheskiy institut im.L.Ya.Karpova, Moskva (Moscow
Physical-Chemical Institute imeni L.Ya.Karpov)

SUBMITTED: April 4, 1957
Card 2/2

AUTHORS:

Lyudvig, Ye. B., Cantmakher, A. R.,
Medvedev, S. S., Corresponding Member, Academy of Sciences,
USSR

20-119-1-24/52

TITLE:

Some Peculiarities of the Carbonium-Polymerization of the
Styrene- α -Methylstyrene System (Nekotoryye osobennosti
karboniyevoy polimerizatsii sistemy stirol- α -metilstirol)

PERIODICAL:

Doklady Akademii Nauk SSSR, 1958, Vol. 119, Nr 1, pp. 90-93
(USSR)

ABSTRACT:

The present paper is devoted to the investigation of a common cationic polymerization of the above-mentioned substances. In the earlier first investigation by the authors a rapid acceleration of the styrene-polymerization in the presence of small additions of α -methylstyrene was observed. Now the dependence of the common polymerization on the composition of the initial mixture was investigated, and compositions of corresponding polymers together with the molecular weights determined. The obtained results were compared with the results of the separate polymerization. SnCl_4 in a solution of ethyl chloride at 0°C was used as catalyst. The curves (figure 1) obtained for various initial relations of the

Card 1/4

Some Peculiarities of the Carbonium-Polymerization of the
Styrene- α -Methylstyrene System

20-119-1-24/52

monomers show that in the concentration range from 0,7 to 1,7 Mol for α -methylstyrene (at a total concentration of the monomers of 2,5 Mol/liter) the polymerization under review takes place without an induction period with a gradually decreasing velocity. The composition of the co-polymers with those of the initial mixtures is given in table 1 which shows that the co-polymers are highly enriched with α -methylstyrene as more active component. The product of the copolymerization-constants in this system is less than 1 (references 1,2). The steric effect manifests itself in a separate polymerization of α -methylstyrene which is confirmed by the reduced thermal effect of its polymerization (ref. 3). The molecular weight of the polymers are given in table 2. The kinetic curves of the domain of the additions of small quantities of the more active component (figure 2) are the most interesting. The intensive effect of an increase in polymerization velocity in the course of time is characteristic here. It begins at a certain stage of the polymerization which is characteristic of every given relation of the polymers. The maximum velocity was observed in a relation of 10 Mol styrene: 1 Mol α -methylstyrene. With increasing

Card 2/4

Some Peculiarities of the Carbonium-Polymerization of the
Styrene- α -Methylstyrene System

20-119-1-24/52

concentration of the catalyst the acceleration of the polymerization increases and its depth decreases (figure 2). The temperature drop highly reduces the relation of the maximum velocity to the initial velocity and lengthens the induction period. The observed maximum velocities exceed the sum of the velocities of the separate polymerization of the same monomers with the same initial concentrations. From these data follows that the effect of the acceleration is connected with the initiation acts. This is also indicated by a powerful influence of water and HCl upon the acceleration: as is to be seen from table 3 the acceleration effect is highly inhibited by these additions. The possible causes of a small acceleration at high HCl-concentrations are discussed. Further a partial effect of the reduction of the initial velocity takes place on the introduction of small additions of α -methylstyrene (figure 4). Such a phenomenon is well known in the radical polymerization. With the exhaustion of the quantity of α -methylstyrene the polymerization velocity shall somewhat increase and approach that of pure styrene. It was proved in this paper that the reactivity of the α -methylstyrene molecule in proportion to the carbonium ion

Card 3/4

Some Peculiarities of the Carbonium-Polymerization of the
Styrene- α -Methylstyrene System

20-119-1-24/52

of styrene is considerably higher than that of styrene.
Small quantities of the first-mentioned more active
component intensively accelerate the styrene polymerization.
This effect apparently is of general nature and is connected
with a great difference of the activity of the monomers.
There are 4 figures, 2 tables, and 3 references, 1 of which
is Soviet.

SUBMITTED: October 9, 1957

Card 4/4

ZABOLOTSKAYA, Ye.V.; GANTMAKHER, A.R.; MEDVEDEV, S.S.

Polymerization of styrene with the simultaneous action of a catalyst
and light. Vysokom. soed. 1 no.3:460-465 Mr '59. (MIRA 12:10)

1. Fiziko-khimicheskiy institut im. L. Ya. Karpova.
(Polymerization) (Styrene)

SPIRIN, Yu.L.; GANTMAKHER, A.R.; MEDVEDEV, S.S.

Mechanism of polymerisation in the presence of alkali metal
organic compounds. Vysokom.soced. 1 no.8:1258-1265
Ag '59. (MIRA 13:2)

1. Fiziko-khimicheskii institut im. L.Ya.Karpova.
(Polymerization) (Alkali metal compounds)

LYUDVIG, Ye.B.; GANTMAKHER, A.R.; MEDVEDEV, S.S.

Characteristics of the mechanism of cationic polymerization.
Part 1: Copolymerization of the systems α -methylstyrene - styrene,
isobutylene - styrene and n-butyl vinyl ether - styrene. Vysokom. soed.
1 no.9:1333-1341 S '59. (MIRA 13:3)

1. Fiziko-khimicheskiy institut im. L.Ya. Karpova.
(Styrene) (Ether) (Propene) (Polymerization)

LYUDVIG, Ye. V.; GANTMAKHER, A.R.; MEDVEDEV, S.S.

Characteristics of the mechanism of cationic polymerization. Part 2:
Mechanism of the fundamental reactions of cationic polymerization.
Vysokom. soed. 1 no.9:1342-1350 S '59. (MIRA 13:3)

1. Fiziko-khimicheskiy institut im. L. Ya. Karpova.
(Polymerization) (Styrene) (Propene)

GANTMAKHER, A.R.; SPIRIN, Yu.L.; MEDVEDEV, S.S.

Polymerization and copolymerization of fluorinated styrenes.
Vysokom.soced. 1 no.10:1526-1530 0 '59. (MIRA 13:3)

1. Fiziko-khimicheskiy institut im.L.Ya.Karpova.
(Styrene) (Polymerization)

5 (2)

AUTHORS:

Gantmakher, A. R., Medvedev, S. S.,
Academician, Lyudvig, Ye. B.

30V/20-127-1-26/65

TITLE:

On the Initiation Mechanism of Cationic Polymerization in the
Presence of Metal Halides (K voprosu o mekhanizme initsii-
rovaniya kationnoy polimerizatsii v prisutsvii galogenidov
metallov)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 127, Nr 1, pp 100 - 103
(USSR)

ABSTRACT:

There are two interpretations concerning the initiation mechanism of carbonium polymerization and of the formation of the primary carbonium ion, respectively: (a) Cationic polymerization cannot proceed in the presence of metal halides without an addition of various co-catalysts. (b) This polymerization is possible under certain conditions without the additions mentioned. The problem of the nature of the co-catalytic additions developed considerably with the progress of investigations. In references 1-3 it was detected for isobutylene polymerization in liquid and in hexane solution that no polymerization takes place without proton-containing additions. Therefrom it results that always proton-containing acids of the type $HB \cdot PX_n$ act as

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On the Initiation Mechanism of Cationic Polymerization in the Presence of Metal Halides SOV/20-127-1-26/65

initiators of the aforesaid polymerization (HB - co-catalyst PX_n -metal halide). This held in the case of low temperatures and media with a low dielectric constant. The results of isoprene- and styrene polymerization in the presence of $SnCl_4$ obtained by the authors showed, however, that the polymerization mentioned proceeds as well without additions at higher temperature and a higher dielectric constant (Ref 4). This fact concerning halogen alkyls and dichloro-ethane without additions (Refs 7,8,11) was confirmed by references 5,6. The authors of the two last-mentioned papers were, however, of the opinion that the solvent plays here the role of a co-catalyst (see Scheme). The scheme mentioned shows that the breaking of the chain in chloro-ethyl and dichloro-ethane with $TiCl_4$ or $SnCl_4$ as catalysts should not depend on the question as to whether the reaction is carried out in the presence of HCl or without this acid, since the carbonium ion is in either case in the field of one and the same compensating ion. This is in contrast to the authors' results according to which HCl additions reduce the molecular weight of polymers produced by the polymerization in halogen alkylic and other solvents (Ref 7). Only the

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molecules bound to the catalysts are effective. From these and other results (Refs 8-11) the authors drew the conclusion that the cationic polymerization may proceed under the direct effect of aprotic acids in halogen alkylic solvents without the participation of specific co-catalysts. This holds also for water (Refs 5,6). In reference 12 it is, however, not denied that both (a) and (b) polymerization methods are possible. The initiation reaction in the monomer - catalyst system proceeds apparently by way of the formation of a π -complex of the catalyst with the monomer. The initiation reaction is caused by an interaction between this complex and the monomer (Refs 8,13; analogy in reference 14). Thus, complex formation effects (Refs 11, 15) are inhibited by additions of H_2O and HCl (Ref 11) which form themselves stable complexes with $SnCl_4$. α -methyl styrene forms complexes with $SnCl_4$ better than the styrene used in references 1-3. It is rather probable that the co-catalysts form in non-polar solvents not only complexes with the catalyst, but also favor the formation of an ion couple by the solvation of the complex. The additions themselves may play this role as

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well as their complexes with the catalyst. Thus, polymerization without co-catalyst is in several systems one of the special cases of the complex nature of the initiation process. There are 16 references, 9 of which are Soviet.

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TITLE: The Copolymerization of Parachlorostyrene With α -Methylstyrene
and Styrene Under the Influence of Alkaline Catalysts

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ABSTRACT: An investigation is made of the reactivity of chlorine-contain-
ing monomers under the influence of lithium-organic and sodium-
organic catalysts in different media. The polymerization took
place in a vacuum, to exclude the effects of air and humidity.
Previous experiments with monomers containing a relatively mo-
bile chlorine atom (chlorovinyl, chloroprene) showed that the
chlorine atom quickly reacts with lithiummethyl and that no poly-
merization takes place even at low temperatures. The chlorine
atom of parachlorostyrene is, however, less mobile, and thus it
is possible to carry out the polymerization. Table 1 gives the
results of the experiments. Lithiummethyl, sodium triphenylmethyl,
 α -sodium naphthalene and γ -radiation were used as catalysts; the
solvents were benzene, ether, triethylamine, and tetrahydrofuran.
The composition of the copolymers strongly depended on the kind

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of catalyst and medium. In the case of hydrocarbons the radical polymerization is predominant in the polymerization with lithium-ethyl. The same radical polymerization also prevails under the effect of γ -radiation. In the case of triethylamine the polymerization according to the anion mechanism is most frequent, while when lithiummethyl is used in ether radical mechanism and anion mechanism are found side by side. The constants of anionic copolymerization calculated for styrene ($\alpha = 0.1 \pm 0.1$) and parachlorostyrene ($\beta = 6.5 \pm 0.1$) show that the introduction of the chlorine atom into styrene increases the activity of the monomer for anionic polymerization. There are 1 table and 2 references, 1 of which is Soviet.

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